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Spectroscopic studies of neutral and chemically oxidized species of β -carotene, lycopene and norbixin in CH_2Cl_2 : Fluorescence from intermediate compounds



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ABSTRACT

Radical cations, dications and oxidized intermediate species of three carotenoids, namely, β -carotene, lycopene and norbixin, were generated in CH₂Cl₂ solutions *via* chemical oxidation using anhydrous FeCl₃. UV–vis, fluorescence and fluorescence-excitation spectroscopic studies were performed to understand and compare the nature of intermediate species generated during the chemical oxidation process and subsequent degradation. The intense emission observed at 550 nm can be assigned to the $S_2 \rightarrow S_0$ ($1^1B_u \rightarrow 1^1A_g$) transition of the carotenoid molecules. The 350 nm excitation during the oxidation process for β -carotene, lycopene and norbixin exhibit intense fluorescence peaks at 492 nm, 493 nm and 500 nm, respectively. These peaks are assigned to intermediate peroxy/epoxy compounds of the three molecules that are formed with molecular oxygen prior to the formation of oxidized short-chain stable compounds.

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