SUPPRESSION OF RECOMBINATION CHANNELS OF DYE–SENSITIZED SOLAR CELLS BY INCORPORATING NANO-PARTICLES OF METALS AND INSULATORS TO THE SEMICONDUCTOR FILM

By

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DECLARATION

"The work described in this thesis was carried out by me under the supervision of Prof. K. Tennakone, Director (former), Institute of Fundamental Studies, Hanthana Road, Kandy and Prof. Pradeep Jayaweera, Professor in Chemistry, Department of Chemistry, University of Sri Jayewardenepura, Nugegoda and a report on this has not been submitted in whole or in part to any university or any other institute for another Degree/Diploma".

Larant

Signature of the candidate

"We certify that the above statement made by the candidate is true and that thesis is suitable for submission to the University for the purpose of evaluation".

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ABSTRACT

In the conversion of solar energy into electricity, devices made of high band gap semiconductor materials are more stable comparable to those made of low band gap semiconductor materials. However, under these circumstances, an efficient absorber material is very essential. In this work, particular dye (Indoline) as a sensitizer and high band gap semiconductors such as TiO₂ and SnO₂ were utilized in the construction of dye–sensitized solar cells.

The performance of dye-sensitized solar cells based on SnO_2 film is inferior due to existing of shallow trap levels belong to it. On the other hand, the existing of these defects on TiO₂ film is minimum and hence, improvement of cell performance based on composite systems becomes marginal for TiO₂ semiconductor. However, the efficiency of indoline–149 sensitized photoelectrochemical solar cells increases significantly when the nanocrystalline TiO₂ is doped with Cu to a concentration similar to the dye concentration. Indoline–149 possesses sulfur in rhodanine rings in addition to carboxylate ligand. Therefore, sulfur in rhodanine rings of indoline–149 too could coordinate with copper in TiO₂ in addition to the anchoring of this dye to TiO₂ surface via the carboxylate ligand. And hence it suppresses the aggregation of the dye molecules. The firm bonding of the indoline–149 molecule to the TiO₂ surface at two points covers TiO_2 surface without leaving any voids. Therefore, suppression of recombination through these voids may increase both photovoltage as well as photocurrent.

Depositing nano-particles of gold on SnO₂ particles showed enhancement of both photovoltage as well as photocurrent. As many articles in literature claims, the surface plasmon resonance of the gold nano-particles deposited on semiconductors enhances the performance of the dye-sensitized solar cells accompanied by an increment of photocurrent. Since the above device improves photovoltage as well, it is realized that in addition to the surface plasmon resonance effect, there may be several factors which dominates the mechanism of this device. The Fermi level of SnO₂ particles rises when it is in contact with a metal having lower work function. Since the particles are in nano-range, to achieve the equalization of Fermi levels of gold and SnO₂ particles, the potential of the conduction band edge of SnO₂ rises up together with the shallow traps. This will result to suppress the recombination of germinated electrons with acceptors in electrolyte which increases the photocurrent. The shift of the conduction band edge to a higher level attributes to higher photovoltage as well.

ZrO₂ is an insulating material with its energy gap in between 5–7 eV. Suppression of recombination processes through the incorporation of the insulating material, ZrO₂ improved cell performance significantly. Many articles in literature revealed that the enhancement of the photocurrent is impossible in coupled type heterostructures formed between insulator and semiconductor. And therefore, the formation of ZrO₂ shell around the SnO₂ crystallites in the [SnO₂] ZrO₂ (capped type heterostructures) can not be ruled out in this regard. This ZrO₂ shell around SnO₂ prevents the recombination of germinated electrons with acceptors in electrolyte. And accumulation of these electrons in SnO₂ particles rises up quasi–Fermi level (QFL) of the [SnO₂] ZrO₂ composite. This will attributes to higher photovoltage and photocurrent of the cell.